2-(4-Methylbenzenesulfonamido)-2-phenylacetic acid

Irfana Mariam,[a] Mehmet Akkurt,[b]* Shahzad Sharif,[a] Noreen Akhtar[a] and Islam Ullah Khan[a]*

[a]Materials Chemistry Laboratory, Department of Chemistry, Government College University, Lahore 54000, Pakistan, and [b]Department of Physics, Faculty of Arts and Sciences, Erciyes University, 38039 Kayseri, Turkey

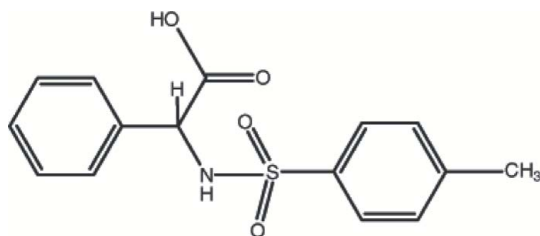Correspondence e-mail: akkurt@erciyes.edu.tr, iukhan.gcu@gmail.com

In the title compound, $C_{15}H_{15}NO_4S$, the dihedral angle between the phenyl and benzene rings is 46.0 (3)° and a weak intramolecular N—H···O interaction is present. The crystal structure is stabilized by intermolecular O—H···O, N—H···O and C—H···O hydrogen bonds.

Related literature

For previous studies on the synthesis of sulfonamide derivatives with phenyl glycine, see: Asiri *et al.* (2009); Arshad *et al.* (2009). For reference structural data, see: Allen *et al.* (1987).

Experimental

Crystal data

$C_{15}H_{15}NO_4S$	V = 1480.5 (5) Å^3
M_r = 305.35	Z = 4
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
a = 5.6592 (12) Å	μ = 0.23 mm^{-1}
b = 11.208 (2) Å	T = 296 K
c = 23.342 (4) Å	0.35 × 0.22 × 0.10 mm

Data collection

Bruker Kappa APEXII CCD diffractometer	(*XABS2*; Parkin *et al.*, 1995) T_{min} = 0.923, T_{max} = 0.977
Absorption correction: refined from ΔF	3753 measured reflections
	3753 independent reflections
	1502 reflections with $I > 2\sigma(I)$

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.066	$\Delta\rho_{max}$ = 0.30 e Å^{-3}
$wR(F^2)$ = 0.192	$\Delta\rho_{min}$ = −0.38 e Å^{-3}
S = 0.94	Absolute structure: Flack (1983),
3753 reflections	1550 Freidel pairs
195 parameters	Flack parameter: −0.11 (19)
H atoms treated by a mixture of independent and constrained refinement	

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O1—H$O1$···O2^i	0.82	1.85	2.655 (6)	168
N1—H$N1$···O1^{ii}	0.85 (5)	2.47 (5)	3.251 (6)	154 (5)
N1—H$N1$···O2	0.85 (5)	2.43 (5)	2.748 (6)	103 (4)
C7—H7···O3^{iii}	0.98	2.43	3.343 (7)	155

Symmetry codes: (i) $x + \frac{1}{2}, -y - \frac{1}{2}, -z + 2$; (ii) $x - 1, y, z$; (iii) $x + 1, y, z$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5142).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.

Arshad, M. N., Tahir, M. N., Khan, I. U., Shafiq, M. & Ahmad, S. (2009). *Acta Cryst.* E**65**, o940.

Asiri, A. M., Akkurt, M., Khan, S. A., Arshad, M. N., Khan, I. U. & Sharif, H. M. A. (2009). *Acta Cryst.* E**65**, o1246–o1247.

Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.

Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.

Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.

Flack, H. D. (1983). *Acta Cryst.* A**39**, 876–881.

Parkin, S., Moezzi, B. & Hope, H. (1995). *J. Appl. Cryst.* **28**, 53–56.

Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.

Spek, A. L. (2009). *Acta Cryst.* D**65**, 148–155.

supplementary materials

2-(4-Methylbenzenesulfonamido)-2-phenylacetic acid

I. Mariam, M. Akkurt, S. Sharif, N. Akhtar and I. U. Khan

Comment

(*R*)-alpha-Amino-benzeneacetic is a side chain component of Ampicillin, Cephalexin and Cephaclor. Cephalexin has *D*-phenylglycyl group as a substituent at the 7-amino position and an unsubstituted methyl group at the 3-position. This is in connection with our previous study on synthesis of sulfonamide derivatives with phenyl glycine (Arshad *et al.*, 2009).

In the title molecule (I), (Fig. 1), bond lengths (Allen *et al.*, 1987) and bond angles are in the range of expected values. The planes of the phenyl and benzene rings (C1–C6) and (C9–C14) make a dihedral angle of 46.0 (3) ° with each other.

In the structure, the adjacent molecules are connected by intermolecular O—H···O, N—H···O and C—H···O hydrogen bonds (Table 1). In Fig. 2, the packing and hydrogen bonding of (I) are shown viewed down *a* axis.

Experimental

Phenyl glycine (1.0 g, 6.6 mmol) was dissolved in 20 ml distilled in a round bottom flask (100 ml). 1*M* Na$_2$CO$_3$ solution was used to maintain pH at 8–9. Para-toluene sulfonyl chloride (1.26 g, 6.6 mmol) was added to the solution, and stirred at room temperature until the para-toluene sulfonylchloride was consumed. On completion of the reaction, while vigorous stirring pH was adjusted 1–2, using 1 *M* HCl. The precipitate formed in this way was filtered off, washed with distilled water, dried and recrystalized in methanol and ethyl acetate (50:50 *v/v*) to yield light brown prisms of (I).

Refinement

The NH H atom was localized from the difference-Fourier map and its coordinates were refined freely. The isotropic temperature parameters of the H atom were calculated as 1.2U_{eq} of the parent atom. H atoms were located geometrically and treated as riding with C—H = 0.98 Å (methine), C—H = 0.96 Å (methyl), C—H = 0.93 Å (aromatic) and O—H = 0.82 Å (hydroxyl) with U_{iso}(H) = 1.2 or 1.5U_{eq}(C, O).
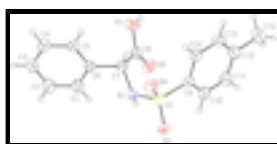
Figures

Fig. 1. An *ORTEP-3* view of the title molecule with the atom numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 30% probability level.
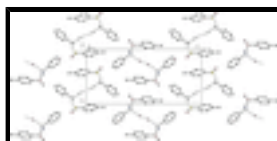

Fig. 2. The packing and hydrogen bonding of the title compound viewed down *a* axis. Hydrogen atoms not involved in the showed interactions have been omitted for clarity.

supplementary materials

2-(4-Methylbenzenesulfonamido)-2-phenylacetic acid

Crystal data

$C_{15}H_{15}NO_4S$	$F_{000} = 640$
$M_r = 305.35$	$D_x = 1.370$ Mg m^{-3}
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: P 2ac 2ab	Cell parameters from 533 reflections
$a = 5.6592$ (12) Å	$\theta = 2.5–15.0º$
$b = 11.208$ (2) Å	$\mu = 0.23$ mm^{-1}
$c = 23.342$ (4) Å	$T = 296$ K
$V = 1480.5$ (5) Å^3	Prism, light brown
$Z = 4$	$0.35 \times 0.22 \times 0.10$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer	3753 independent reflections
Radiation source: sealed tube	1502 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{int} = 0.0000$
$T = 296$ K	$\theta_{max} = 28.6º$
φ and ω scans	$\theta_{min} = 1.7º$
Absorption correction: part of the refinement model (ΔF) (XABS2; Parkin *et al.*, 1995)	$h = -7 \rightarrow 7$
$T_{min} = 0.923$, $T_{max} = 0.977$	$k = 0 \rightarrow 15$
3753 measured reflections	$l = 0 \rightarrow 31$

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H atoms treated by a mixture of independent and constrained refinement
$R[F^2 > 2\sigma(F^2)] = 0.066$	$w = 1/[\sigma^2(F_o{}^2) + (0.0657P)^2]$ where $P = (F_o{}^2 + 2F_c{}^2)/3$
$wR(F^2) = 0.192$	$(\Delta/\sigma)_{max} < 0.001$
$S = 0.94$	$\Delta\rho_{max} = 0.30$ e Å^{-3}
3753 reflections	$\Delta\rho_{min} = -0.38$ e Å^{-3}
195 parameters	Extinction correction: none
Primary atom site location: structure-invariant direct methods	Absolute structure: Flack (1983), 1550 Freidel pairs
Secondary atom site location: difference Fourier map	Flack parameter: −0.11 (19)

Special details

Experimental. Absorption correction: XABS2; Parkin *et al.* (1995), linear fit to sin(theta)/lambda - 12 parameters

Geometry. Bond distances, angles *etc*. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R-factors wR and all goodnesses of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating -R-factor-obs *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
S1	0.0144 (3)	0.06971 (13)	0.90449 (6)	0.0506 (5)
O1	0.6089 (8)	−0.1884 (4)	0.93717 (17)	0.0620 (16)
O2	0.2309 (8)	−0.2174 (4)	0.96274 (18)	0.0683 (17)
O3	−0.2356 (6)	0.0816 (4)	0.90143 (16)	0.0590 (16)
O4	0.1628 (7)	0.1504 (4)	0.87441 (17)	0.0607 (17)
N1	0.0702 (7)	−0.0631 (4)	0.87997 (19)	0.0470 (17)
C1	0.2019 (12)	−0.2702 (6)	0.8097 (3)	0.070 (3)
C2	0.2455 (14)	−0.3397 (7)	0.7628 (3)	0.085 (3)
C3	0.4491 (12)	−0.3245 (6)	0.7318 (3)	0.073 (3)
C4	0.6069 (13)	−0.2379 (6)	0.7467 (3)	0.072 (3)
C5	0.5634 (9)	−0.1673 (6)	0.7937 (3)	0.061 (3)
C6	0.3604 (10)	−0.1822 (5)	0.8253 (2)	0.0463 (19)
C7	0.3143 (9)	−0.1052 (5)	0.8779 (2)	0.0470 (19)
C8	0.3763 (11)	−0.1745 (5)	0.9306 (2)	0.050 (2)
C9	0.0950 (9)	0.0722 (5)	0.9766 (2)	0.0490 (19)
C10	−0.0428 (11)	0.0200 (5)	1.0177 (3)	0.062 (2)
C11	0.0257 (12)	0.0163 (6)	1.0734 (3)	0.070 (3)
C12	0.2384 (12)	0.0650 (6)	1.0907 (3)	0.067 (2)
C13	0.3774 (11)	0.1171 (6)	1.0498 (3)	0.065 (3)
C14	0.3091 (10)	0.1222 (5)	0.9926 (3)	0.057 (2)
C15	0.3157 (14)	0.0600 (8)	1.1524 (3)	0.099 (3)
H1	0.06550	−0.28220	0.83110	0.0840*
HO1	0.63490	−0.22620	0.96660	0.0930*
H2	0.13660	−0.39750	0.75180	0.1010*
HN1	−0.020 (9)	−0.116 (5)	0.894 (2)	0.0560*
H3	0.47970	−0.37340	0.70040	0.0880*
H4	0.74340	−0.22670	0.72520	0.0870*
H5	0.67200	−0.10900	0.80420	0.0730*
H7	0.41830	−0.03540	0.87600	0.0560*
H10	−0.18660	−0.01380	1.00720	0.0740*
H11	−0.07210	−0.01950	1.10040	0.0840*
H13	0.52160	0.15010	1.06050	0.0780*
H14	0.40570	0.15860	0.96550	0.0690*
H15A	0.48070	0.04040	1.15420	0.1490*
H15B	0.22600	0.00020	1.17220	0.1490*

H15C	0.28970	0.13620	1.17010	0.1490*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0449 (8)	0.0563 (8)	0.0505 (8)	0.0024 (8)	0.0026 (8)	0.0009 (8)
O1	0.051 (2)	0.074 (3)	0.061 (3)	−0.001 (2)	−0.012 (2)	0.021 (2)
O2	0.066 (3)	0.082 (3)	0.057 (3)	0.003 (2)	0.016 (2)	0.017 (2)
O3	0.034 (2)	0.076 (3)	0.067 (3)	0.0123 (19)	−0.0037 (19)	−0.002 (2)
O4	0.059 (3)	0.055 (3)	0.068 (3)	−0.002 (2)	0.013 (2)	0.012 (2)
N1	0.040 (3)	0.054 (3)	0.047 (3)	0.003 (2)	0.001 (2)	−0.002 (2)
C1	0.070 (4)	0.078 (5)	0.062 (4)	−0.014 (4)	0.008 (4)	−0.024 (4)
C2	0.082 (5)	0.083 (6)	0.089 (5)	−0.015 (4)	0.001 (4)	−0.034 (4)
C3	0.065 (5)	0.095 (6)	0.060 (4)	0.016 (4)	0.001 (3)	−0.022 (4)
C4	0.062 (4)	0.096 (6)	0.059 (5)	0.007 (4)	0.011 (4)	−0.009 (4)
C5	0.044 (4)	0.085 (5)	0.053 (4)	0.002 (3)	0.003 (3)	−0.009 (3)
C6	0.048 (3)	0.052 (4)	0.039 (3)	0.003 (3)	0.000 (3)	0.000 (3)
C7	0.038 (3)	0.057 (4)	0.046 (3)	−0.004 (2)	−0.002 (2)	0.001 (3)
C8	0.053 (4)	0.057 (4)	0.040 (3)	−0.008 (3)	0.002 (3)	−0.003 (3)
C9	0.038 (3)	0.056 (3)	0.053 (4)	0.004 (3)	0.004 (3)	−0.007 (3)
C10	0.055 (4)	0.069 (4)	0.062 (4)	−0.008 (3)	0.004 (3)	−0.005 (3)
C11	0.075 (5)	0.084 (5)	0.052 (4)	−0.003 (4)	0.005 (4)	0.000 (3)
C12	0.071 (4)	0.073 (4)	0.056 (4)	0.010 (4)	−0.006 (4)	−0.010 (4)
C13	0.053 (4)	0.069 (4)	0.072 (5)	−0.003 (3)	−0.004 (4)	−0.014 (4)
C14	0.050 (4)	0.062 (4)	0.060 (4)	0.000 (3)	0.003 (3)	−0.007 (3)
C15	0.108 (6)	0.126 (7)	0.064 (5)	0.017 (6)	−0.017 (4)	−0.019 (5)

Geometric parameters (Å, °)

S1—O3	1.423 (4)	C10—C11	1.357 (10)
S1—O4	1.420 (4)	C11—C12	1.382 (10)
S1—N1	1.626 (5)	C12—C13	1.368 (10)
S1—C9	1.744 (5)	C12—C15	1.506 (10)
O1—C8	1.334 (8)	C13—C14	1.391 (10)
O2—C8	1.213 (7)	C1—H1	0.9300
O1—HO1	0.8200	C2—H2	0.9300
N1—C7	1.461 (7)	C3—H3	0.9300
N1—HN1	0.85 (5)	C4—H4	0.9300
C1—C6	1.382 (9)	C5—H5	0.9300
C1—C2	1.366 (10)	C7—H7	0.9800
C2—C3	1.371 (10)	C10—H10	0.9300
C3—C4	1.364 (10)	C11—H11	0.9300
C4—C5	1.375 (10)	C13—H13	0.9300
C5—C6	1.375 (8)	C14—H14	0.9300
C6—C7	1.523 (7)	C15—H15A	0.9600
C7—C8	1.497 (7)	C15—H15B	0.9600
C9—C14	1.386 (8)	C15—H15C	0.9600
C9—C10	1.368 (8)		

O3—S1—O4	120.2 (3)	C13—C12—C15	121.1 (6)
O3—S1—N1	105.1 (2)	C12—C13—C14	121.8 (6)
O3—S1—C9	107.9 (2)	C9—C14—C13	119.0 (6)
O4—S1—N1	107.1 (2)	C2—C1—H1	120.00
O4—S1—C9	108.2 (3)	C6—C1—H1	120.00
N1—S1—C9	107.7 (3)	C1—C2—H2	120.00
C8—O1—HO1	109.00	C3—C2—H2	120.00
S1—N1—C7	119.4 (3)	C2—C3—H3	120.00
S1—N1—HN1	113 (4)	C4—C3—H3	120.00
C7—N1—HN1	111 (4)	C3—C4—H4	120.00
C2—C1—C6	120.1 (6)	C5—C4—H4	120.00
C1—C2—C3	120.3 (7)	C4—C5—H5	120.00
C2—C3—C4	120.3 (7)	C6—C5—H5	120.00
C3—C4—C5	119.7 (6)	N1—C7—H7	108.00
C4—C5—C6	120.5 (6)	C6—C7—H7	108.00
C1—C6—C5	119.2 (5)	C8—C7—H7	108.00
C1—C6—C7	120.4 (5)	C9—C10—H10	119.00
C5—C6—C7	120.4 (5)	C11—C10—H10	119.00
N1—C7—C6	111.8 (4)	C10—C11—H11	120.00
C6—C7—C8	109.2 (4)	C12—C11—H11	119.00
N1—C7—C8	111.2 (4)	C12—C13—H13	119.00
O2—C8—C7	123.7 (5)	C14—C13—H13	119.00
O1—C8—C7	112.7 (5)	C9—C14—H14	121.00
O1—C8—O2	123.5 (5)	C13—C14—H14	120.00
S1—C9—C10	121.4 (4)	C12—C15—H15A	109.00
S1—C9—C14	119.7 (4)	C12—C15—H15B	109.00
C10—C9—C14	118.8 (5)	C12—C15—H15C	110.00
C9—C10—C11	121.5 (6)	H15A—C15—H15B	109.00
C10—C11—C12	121.1 (6)	H15A—C15—H15C	109.00
C11—C12—C15	121.2 (6)	H15B—C15—H15C	109.00
C11—C12—C13	117.8 (6)		
O3—S1—N1—C7	−179.2 (4)	C1—C6—C7—N1	−45.4 (7)
O4—S1—N1—C7	51.8 (4)	C1—C6—C7—C8	78.1 (7)
C9—S1—N1—C7	−64.3 (4)	C5—C6—C7—N1	136.5 (5)
O3—S1—C9—C10	36.7 (6)	C5—C6—C7—C8	−100.0 (6)
O3—S1—C9—C14	−146.9 (5)	N1—C7—C8—O1	−164.1 (4)
O4—S1—C9—C10	168.2 (5)	N1—C7—C8—O2	18.5 (7)
O4—S1—C9—C14	−15.4 (5)	C6—C7—C8—O1	72.0 (6)
N1—S1—C9—C10	−76.3 (5)	C6—C7—C8—O2	−105.3 (6)
N1—S1—C9—C14	100.1 (5)	S1—C9—C10—C11	176.5 (5)
S1—N1—C7—C6	−143.9 (4)	C14—C9—C10—C11	0.0 (9)
S1—N1—C7—C8	93.8 (5)	S1—C9—C14—C13	−176.2 (5)
C6—C1—C2—C3	1.6 (11)	C10—C9—C14—C13	0.4 (8)
C2—C1—C6—C5	−1.2 (10)	C9—C10—C11—C12	−0.2 (10)
C2—C1—C6—C7	−179.4 (6)	C10—C11—C12—C13	0.0 (10)
C1—C2—C3—C4	−1.6 (11)	C10—C11—C12—C15	−179.4 (7)
C2—C3—C4—C5	1.3 (11)	C11—C12—C13—C14	0.3 (10)
C3—C4—C5—C6	−0.9 (10)	C15—C12—C13—C14	179.8 (6)
C4—C5—C6—C1	0.9 (9)	C12—C13—C14—C9	−0.5 (9)

C4—C5—C6—C7	179.1 (6)

Hydrogen-bond geometry (Å, °)

D—H⋯A	D—H	H⋯A	D⋯A	D—H⋯A
O1—HO1⋯O2^i	0.82	1.85	2.655 (6)	168
N1—HN1⋯O1^{ii}	0.85 (5)	2.47 (5)	3.251 (6)	154 (5)
N1—HN1⋯O2	0.85 (5)	2.43 (5)	2.748 (6)	103 (4)
C7—H7⋯O3^{iii}	0.98	2.43	3.343 (7)	155

Symmetry codes: (i) $x+1/2, -y-1/2, -z+2$; (ii) $x-1, y, z$; (iii) $x+1, y, z$.

Fig. 1

Fig. 2